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M.Sc. DEGREE (C.S.S.) EXAMINATION, AUGUST 2015

Second Semester

Faculty of Science

Branch : Chemistry

AN2C07/AP2C07/CH2C07/PH2C07/POH2C07—CHEMICAL BONDING AND COMPUTATIONAL CHEMISTRY

(Common to all branches of Chemistry)

(2012 Admission onwards)

Time: Three Hours

Maximum Weight: 30

Section A

Answer any ten questions. Each question carries a weight of 1.

- 1. What is meant by free valance?
- 2. In what respects the quantum mechanical harmonic oscillator differs from classical oscillator?
- 3. Plot the radial portions of the 4s, 4p, 4d and 4f hydrogen like wave functions.
- 4. Explain the quantum mechanical treatment for Pauli's exclusion principle.
- Calculate the average potential and kinetic energies of a harmonic oscillator in the ground state.Compare the results obtained with the classical results.
- 6. What are Slater determinants?
- Calculate the π-bond order for the excited state of butadiene.
- Write down the Hamiltonian operator for electrons in hydrogen molecule and explain the terms involved.
- 9. Write down the Hülckel determinants for benzene and butadiene.
- 10. What are the transformation properties of atomic orbitals?
- 11. What is meant by Configuration Interaction?
- 12. What are the basic principles of Maller Plesset Perturbation Theory?
- 13. What is meant by Koopman's theorem? What are its applications?

 $(10 \times 1 = 10)$

Section B

Answer any five questions. Each question carries a weight of 2.

- 14. Discuss the qualitative idea of Hellmann-Feynman Theorem.
- 15. Explain the quantum mechanical treatment of sp and sp^2 hybridizations.
- 16. Deduce the Hückel molecular orbitals for ethylene.
- 17. Give the systematic procedure for the determination of point groups.
- 18. Draw the MO diagrams of H2O and NH3.
- 19. Explain the Density Functional Theorem.
- 20. Discuss the general format of GAMESS.
- 21. Discuss the Hartee-Fock method.

 $(5 \times 2 = 10)$

Section C

Answer any two questions, Each question carries a weight of 5.

- 22. What are Hermite polynomials? How are they used in solving the Schrödinger equation for a harmonic oscillator?
- 23. Construct the ground state wave function for H_2 molecule according to VB method. Compare it with MO wave function.
- 24. Deduce the hybrid orbitals of BF3 and PCl5 molecules using group theoretical treatment.
- 25. What are the applications of Computation chemistry?

 $(2 \times 5 = 10)$