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Name.....

# M.Sc. DEGREE (CSS) EXAMINATION, MARCH 2013

### First Semester

Faculty of Science

Branch: Chemistry

AN 1C 02 / AP 1C 02 / CH 1C 02 / PH 1C 02 / PO H1C 02—STRUCTURAL AND MOLECULAR ORGANIC CHEMISTRY

(Common to All Branches of Chemistry)

[2012 Admissions]

Time: Three Hours

Maximum Weight: 30

### Section A

Answer any ten questions. Each questions carries a weight of 1.

- Draw the line diagram of hept-4, 6-dien-2-yne. Label each carbon and predict the hybridisation and geometry of each carbon centre.
- 2. Which is more acidic and why-Chloroacetic acid and acetic acid.
- 3. Explain aromatic character of cyclo pentadienyl anion using Hückel rule.
- 4. How NMR is sued as a tool for aromaticity?
- 5. What is primary kinetic isotopic effect?
- 6. Write two example for Hard and Soft acids and bases.
- Explain the mechanism of ester hydrolysis by B<sub>AC</sub>2 mechanism.
- 8. Give two method for the determination of geometrical isomers. Explain.
- 9. Assign E, Z nomenclature of following :-

10. What is atropisomerism?

11. What is the significance of Curtin Hammett principle?

12. Draw the conformers of cis 1, 4-di-t-butyl cyclohexane. Explain their stabilities.

13. Draw and explain the stereochemical relation for the formation of 2-butene from 2-bromo-butane through  ${\rm E}_2$  mechanism.

 $(10 \times 1 = 10)$ 

## Section B

Answer five questions by attempting not more than three questions from each bunch.

Each questions carries a weight of 2.

## BUNCH I (Problem Type)

14. Predicts the products and explain the mechanism of following :-

15. Complete the following reaction.

16. (a) Assign R or S configuration of following :-

(b) Draw the R, S configuration of trans-Cyclo-octane.

17. Predicts the products and explain the mechanism of the following :-

$$\begin{array}{c} \operatorname{CH_3} \\ | \\ \operatorname{C_2H_5\overline{O}} + \operatorname{CH_3} - \operatorname{C} - \operatorname{CH_2CH_3} \end{array} \longrightarrow \\ | \\ \operatorname{Cl} \end{array}$$

### Bunch II (Short Essay Type)

- 18. Write a note on Wheland intermediate. Give the evidence for its formation.
- 19. What is Barton Reaction. Discuss its mechanism?
- 20. What is Enantiotopic and diastereotopic ligands? How NMR used to distinct it?
- 21. Discuss the stereochemistry and stability of cis and trans 1, 2-dimethyl cyclohexane.

 $(5 \times 2 = 10)$ 

#### Section C

Answer any two questions.

Each question carries a weight of 5.

- 22. (a) Explain aromaticity using HMO theory.
  - (b) Define the terms aromatic, anti-aromatic and non-aromatic using suitable examples.
  - (c) Briefly explain the S<sub>N</sub>1 % S<sub>N</sub>2 mechanism.
- 23. (a) Briefly comments on Hammett equation and linear free energy relationship.
  - (b) Give a detailed account of A<sub>AC</sub>2 and B<sub>AL</sub>1 mechanism of ester hydrolysis with experimental evidences.
- 24. Explain with suitable examples to the stereochemistry of C, N, S based chiral centres.
- Illustrate with suitable examples to the conformational analysis of Fused and bridged bicyclic system.

 $(2 \times 5 = 10)$