



QP CODE: 19101048

Reg No	:	
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B.Sc.DEGREE (CBCS) EXAMINATION, DECEMBER 2018

First Semester

B.Sc Food Science & Quality Control Model III

Core Course - FS1CRT02 - BASIC FOOD CHEMISTRY

2017 Admission (Reappearance)

BC6D7325

Maximum Marks: 80 Time: 3 Hours

Part A

Answer any ten questions.

Each question carries 2 marks.

- 1. Justify the statement that water is found to melt and boil at unusually high temperatures.
- 2. List the water soluble vitamins.
- 3. Draw the straight and ring structure of glucose.
- 4. Draw the structure of sucrose.
- 5. Illustrate the reaction of fermentation fo glucose.
- 6. Explain the principle of electrophoretic separation of proteins.
- 7. Distinguish between salting in and salting out effect of protein.
- 8. Discuss on the specificity of enzyme catalyzed reactions.
- 9. Define compound lipids with example.
- 10. Explain the synergism in antioxidant activity with examples.
- 11. Name the pigment present in tomato and anatto.
- 12. Discuss on flavonoids.

 $(10 \times 2 = 20)$

Part B

Answer any **six** questions.

Each question carries 5 marks.

- 13. Distinguish between amylose and amylopectin.
- 14. Explain Maillard reaction and its importance.
- 15. Explain the reactions of aminoacids with formaldehyde and nitrous acid.



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- 16. Explain the mechanism of non competitive inhibition.
- 17. Explain the regulation of enzyme activity.
- 18. Discuss any five physical properties of lipids.
- 19. Explain R M P K values with its significance.
- 20. Explain hydrogenation of fat.
- 21. Distinguish between rancidity and reversion.

 $(6 \times 5 = 30)$

Part C

Answer any two questions.

Each question carries 15 marks.

- 22. Explain the technology of oil and fat processing.
- 23. Explain the structure of protein.
- 24. Describe the application of enzymes in food industry.
- 25. Explain the mechanism of rancidity with factors affecting it and the methods of measurement.

 $(2 \times 15 = 30)$

